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## **Palladium-Catalyzed Reaction of Acylzirconocene Chloride and Stereoselective Formation of Bicyclo[3.3.0] Compounds**

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## **ABSTRACT**



The acylzirconocene chloride complex as an acyl group donor reacts with ω-unsaturated α,*β*-enones and -ynones under Pd-Me<sub>2</sub>Zn(Me<sub>2</sub>AlCl)**catalyzed conditions to give stereoselectively bicyclo[3.3.0] compounds through (i) formation of a Pd(II) intermediate by an oxidative addition of the Pd(0) catalyst to an enone function, (ii) cyclization of the Pd intermediate to an** *ω***-unsaturated group, (iii) an acyl group transfer from zirconium to Pd metal, (iv) reductive elimination of the Pd metal, and (v) intramolecular cis-selective aldol reaction.**

The search for new reactivity of stable acylzirconocene chloride complexes **1**, which can be easily prepared through the hydrozirconation of unsaturated compounds and subsequent carbon monoxide insertion, $\frac{1}{1}$  is the focus of our current investigation.2 Recently, we reported regioselective access to  $\alpha$ -ketol and 1,4-diketone compounds through Pd(0)catalyzed reactions of 1 with  $\alpha$ , $\beta$ -unsaturated enone derivatives  $2$  (Scheme 1).<sup>2d,e</sup> In these reactions, we postulated a series of reactions: the formation of Pd(II) intermediate species **A** generated by an electron transfer from a Pd(0) catalyst to  $\alpha$ , $\beta$ -enones, the transfer of an acyl group from 1 to Pd(II) intermediate **A** (transmetalation), and the reductive elimination of Pd(0) giving acylation product. The regioselectivity of the reaction was efficiently controlled by the

choice of additives; that is, an addition of Lewis acid or phosphine ligand to the reaction mixture afforded 1,4 addition or 1,2-addition product, respectively.

**Scheme 1.** Pd-Catalyzed Regioselective Addition of Acylzirconocene Chloride to  $\alpha$ , $\beta$ -Enones



As a working hypothesis, when  $\omega$ -unsaturated  $\alpha$ , $\beta$ -enone compound **3** was treated with **1** under identical conditions to that of  $\alpha$ , $\beta$ -enone 2, one would expect cyclization-

<sup>(1)</sup> Bertelo, C. A.; Schwartz*,* J. *J. Am. Chem. Soc*., **1975**, *97*, 228.

<sup>(2) (</sup>a) Hanzawa, Y.; Kakuuchi. A.; Yabe, M.; Narita, K.; Tabuchi, N.; Taguchi, T. *Tetrahedron Lett*. **2001**, *42*, 1737. (b) Hanzawa, Y.; Narita, K.; Taguchi, T. *Tetrahedron Lett*. **2000**, *41*, 109. (c) Kakuuchi, A.; Taguchi, T.; Hanzawa, Y. *Tetrahedron Lett*. **2001**, *42*, 1547. (d) Hanzawa, Y.; Tabuchi, N.; Saito, K.; Noguchi, S.; Taguchi, T. *Angew*. *Chem., Int. Ed*. **1999**, *38*, 2395. (e) Hanzawa, Y.; Tabuchi, N.; Taguchi, T. *Tetrahedron Lett*. **1998**, *39*, 8141. (f) Hanzawa, Y.; Tabuchi, N.; Taguchi, T. *Tetrahedron Lett*. **1998**, *39*, 6249. (g) Harada, S.; Taguchi, T.; Tabuchi, N.; Narita, K.; Hanzawa, Y. *Angew*. *Chem., Int. Ed*. **1998**, *37*, 1696.

acylation to occur through an intramolecular reaction of Pd- (II) intermediate species **B** to the *ω*-unsaturated bond (Scheme 2). In this paper, we describe the stereoselective formation of bicyclo[3.3.0] compound **4** by the Pd-catalyzed cyclization-acylation reaction of acylzirconocene chlorides **1** with  $\omega$ -unsaturated  $\alpha$ , $\beta$ -enone **3** (Scheme 2).



A series of reactions were involved in the conversion from **3** to **4**: (i) cyclization of Pd intermediate **B**, (ii) regioselective acylation, and (iii) subsequent intramolecular aldol reaction of the intermediately formed diketone. Recently, Montgomery et al. reported on the stereoselective formation of bicyclo- [3.3.0] compounds from *ω*-alkynyl-α,*β*-enone compounds through cyclization of the *π*-allylic nickel complex and subsequent reaction with electrophiles.<sup>3</sup> They also reported on the usability of vinylzirconocene chloride as a nucleophilic reagent in a similar type of reaction.<sup>4</sup> In our preliminary experiment, the validity of our hypothesis was confirmed by the use of a  $Pd(OAc)<sub>2</sub>-BF<sub>3</sub>·OEt<sub>2</sub>$  catalyst, which gave the 1,4-addition products in the reaction of acylzirconocene chloride 1 with  $\alpha, \beta$ -enones 2<sup>2</sup> Thus, the reaction of *n*-nonanovlationogene chloride (19, 1.3–2.0 equiv) with *<sup>n</sup>*-nonanoylzirconocene chloride (**1a**, 1.3-2.0 equiv) with compound **3a** (1 equiv) in the presence of 10 mol % Pd-  $(OAc)_2$  and  $BF_3$ · $OEt_2$  (1 equiv) in THF-ether (1:2) at ambient temperature gave mostly a 1,4-addition product together with a small amount (10% yield) of bicyclo[3.3.0] octane derivative **4a** as the sole stereoisomer (Scheme 3).

Catalysts such as  $Ni(COD)_2$ ,  $Ni(acac)_2/DIBAL-H$ ,  $PdCl_2 (PPh_3)$ , and Pd(acac)<sub>2</sub> or the use of a Lewis acid other than  $BF_3$ <sup>\*</sup>OEt<sub>2</sub> as an additive (e.g., TiCl<sub>4</sub>, ZnCl<sub>2</sub>, AlC<sub>3</sub>) did not improve the yield of **4a**. Use of an equivalent amount of dimethylzinc (Me<sub>2</sub>Zn) or dimethylaluminum chloride (Me<sub>2</sub>-AlCl) as an additive, however, significantly increased the yields of **4a** to 51% or 70%, respectively, and the presence of the stereoisomer of **4a** was not observed. The structure of **4a** was determined by NMR experiments (DEPT, H-<sup>H</sup> COSY, HMQC, and NOESY) as shown in Scheme 3, and the structure of **4a** was secured by comparing the NMR data of **4b**, which were obtained in a crystalline form, and the



stereochemistry was confirmed by X-ray analysis.<sup>5</sup> The results of the reactions of **1** (2.0 equiv) with *ω*-unsaturated  $\alpha$ , $\beta$ -enones **3** (1 equiv) under the catalytic conditions [Pd- $(OAc)_2$  (10 mol %)-Me<sub>2</sub>Zn (1 equiv) or -Me<sub>2</sub>AlCl (1 equiv)] are listed in Table 1. The Thorpe-Ingold effect<sup>6</sup> plays a vital role in the formation of  $4$  (entries  $1-5$ ). A heteroatomcontaining substrate can be an efficient reactant, as well. Thus, a heterocyclic bicyclo[3.3.0] compound was obtained in fair yields (entries  $6-9$ ). For the present transformation, acylzirconocene chloride complex **1** is not restricted to saturated acylzirconocene chloride, and  $\alpha$ , $\beta$ -unsaturated acylzirconocene chloride can be used as an acyl group donor, as well (entries 7 and 8). *ω*-Alkynyl substrate **3** also reacted with **1a** to give the acid-labile bicyclo[3.3.0] compound (entry  $10$ ).<sup>7</sup> The structure of product 4 in Table 1 was determined by analogy of the NMR data with that of **4b**. It should be stressed that Montgomery's procedure,<sup>4</sup> which is reported for formation of bicyclo[3.3.0] compounds, is restricted to *<sup>ω</sup>-alkynyl-*R,*â-enone* compounds as a starting material, and the cyclization product could not be obtained in the reaction of *ω-alkenyl*-α,*β-enone* compounds.

It is worth noting that the catalyst generated *in situ* by the reduction of  $Pd(OAc)$ <sub>2</sub> with DIBAL-H (1 equiv) in THFether (1:2) indicates a comparable efficiency with  $Pd(OAc)<sub>2</sub>$ / Me2Zn for the formation of **4**. <sup>8</sup> Attempted cyclization reactions of an ene-yne compound, which has no conjugated enone system, did not give the cyclized product. Thus, the  $\alpha$ , $\beta$ -enone functionality in **3** is necessary for bringing about

<sup>(3) (</sup>a) Montgomery, J. *Acc*. *Chem. Res*. **2000**, *33*, 467. (b) Montgomery, J.; Chowdhury, S. K.; Amarasinghe, K. K. D.; Heeg, M. J. *J. Am. Chem. Soc.* **2000**, *122*, 6775. (c) Montgomery, J.; Oblinger, E.; Savchenko, A. V. *J. Am. Chem. Soc.* **1997**, *119*, 4911.

<sup>(4)</sup> Ni, Y.; Amarasinghe, K. K. D.; Montgomery, *J. Org. Lett.* **2002**, *4*, 1743.

<sup>(5)</sup> Crystal data for **4b**:  $C_{27}H_{34}O_2$ ,  $M = 390.567$ , triclinic, *P*1,  $a = 860(13)$  Å  $b = 12.4020(14)$  Å  $c = 30.931(9)$  Å  $\alpha = 84.576(11)$ <sup>o</sup> *B* 5.8860(13) Å,  $b = 12.4020(14)$  Å,  $c = 30.931(9)$  Å,  $\alpha = 84.576(11)^\circ$ ,  $\beta = 85.879(9)^\circ$ ,  $\gamma = 85.835(9)^\circ$ ,  $V = 2237.1(8)$  Å<sup>3</sup>,  $Z = 4$ ,  $D_x = 1.160$  Mg<br>m<sup>-3</sup> Mo Kα radiation  $\lambda = 0.71073$ ,  $\mu = 0.07$  mm<sup>-1</sup>,  $T = 150$  K m<sup>-3</sup>, Mo Kα radiation,  $λ = 0.71073$ ,  $μ = 0.07$  mm<sup>-1</sup>,  $T = 150$  K.; 7835 independent reflections,  $R(gt) = 0.079$ ,  $wR(gt) = 0.142$  for 3534 observed reflections  $[|F_0| > 1\sigma(|F_0|), 2\theta = 54.26^{\circ}]$  and 524 parameters. CCDC-186221.

<sup>(6)</sup> Beesley, R. M.; Ingold, C. K.;, Thorpe, J. F. *J. Am. Chem. Soc.* **1985**, *107*, 1080.

<sup>(7)</sup> Transposition of the hydroxyl and the double bond was observed during the isolation by silica gel column chromatography, and a similar observation has been made by Montgomery et al.3

<sup>(8)</sup> A reduction of  $Ni (acac)_2$  by 1 equiv of DIBAL-H has been reported to give a low-valent nickel catalyst. (a) Dayrit, F. M.; Schwartz, J. *J. Am. Chem. Soc.* **1981**, *103*, 4466. (b) Schwartz, J.; Loots, M. J.; Kosugi, H. *J. Am. Chem. Soc.* **1980**, *102*, 1333. (c) Dayrit, F. M.; Gladkowski, D. E.; Schwartz, J. *J. Am. Chem. Soc.* **1980**, *102*, 4466.

Table 1. Formation of 4 through Pd(OAc)<sub>2</sub>-Catalyzed Reaction*<sup>a</sup>*

	1	3		4
entry	R	X	$R^1$	(yield %) <sup>b</sup>
1	$nC_8H_{17}$ 1a	CH <sub>2</sub>	Ph	26
2	$nC_8H_{17}$	Me <sub>2</sub> C За	Ph	4a 70° 51 <sup>d</sup>
3	Ph(CH <sub>2</sub> ) <sub>4</sub>	Me <sub>2</sub> C	Ph	$4b$ 52 <sup>d</sup>
4	$nC_8H_{17}$	(BnOCH <sub>2</sub> ) <sub>2</sub> C		$53^{\circ}63^{\circ}$
5	$nC_8H_{17}$	(BnOCH <sub>2</sub> ) <sub>2</sub> C Me		$48^\circ$
6	$nC_8H_{17}$	Ω	Ph	$54^{\circ} 83^{\circ}$
7	$n - C_6 H_{13}$	O	Ph	60 <sup>d</sup>
	8TBDPSO(CH2)2	Ω	Ph	72 <sup>d</sup>
9	$nC_8H_{17}$	BocN	Ph	$38^{\circ}$ 41 <sup>d</sup>
10	$nC_8H_{17}$	COPh		COPh OH $nC_8H_{17}$ 48 <sup>c</sup>

<sup>*a*</sup> All reactions were carried out by the use of Pd(OAc)<sub>2</sub> (10 mol %)– Me<sub>2</sub>AlCl (1 equiv) or Me<sub>2</sub>Zn (1 equiv) in THF/Et<sub>2</sub>O (1:2) at ambient temperature. <sup>b</sup> Isolated yield was calculated from 3. <sup>c</sup> Me<sub>2</sub>AlCl as an additive. *<sup>d</sup>* Me2Zn as an additive.

the present transformation.<sup>9</sup>  $\omega$ -Alkenyl- $\alpha$ , $\beta$ -ynone compound **5** also reacts with **1a** to give unsaturated bicyclo[3.3.0] compounds in fair yields (Table 2).

**Table 2.** Reaction of  $\omega$ -Alkenyl- $\alpha$ , $\beta$ -ynones **5** with **1a**  $\mathsf{COR}^1$ COR' 1a  $Pd(OAc)_2$ ՛*ո-*C<sub>8</sub> H<sub>17</sub>  $Me<sub>2</sub>Zn$ **5** entry X  $R^1$  yield  $(\%)^a$ 1  $Me_2C$  Ph 56 2 O 45 3  $BnO(CH<sub>2</sub>)<sub>2</sub>C$  55 4  $BnO(CH<sub>2</sub>)<sub>2</sub>C$  Me 48 *<sup>a</sup>* Isolated yield based on **5**.

Thus, the formation of bicyclo[3.3.0] compound **4** is a sequential result of (i) formation of Pd(II) complex **B** through an oxidative addition to the enone function from the *in situ* generated Pd(0) catalyst, (ii) cyclization of Pd(II) complex intermediate **B** to the  $\omega$ -unsaturated group, (iii) acyl group transfer from Zr to Pd (transmetalation), (iv) reductive elimination of the Pd(0), and (v) intramolecular aldol reaction (Scheme 4). $^{10}$  It should be mentioned that we cannot



rigorously rule out the possibility of the intervention of an oxapalladacycle complex **C** (Scheme 4) in the present transformation.11 The *cis*-stereochemistry between the hydroxyl and ketone functional groups in products **4** indicates that the intramolecular aldol reactions in the present transformation have to be *cis*-selective. To confirm the cisselectivity of the intramolecular aldol reaction, the reaction of keto-enone compound **<sup>6</sup>** with **1a** was examined under the same catalytic conditions  $[Pd(OAc)<sub>2</sub>(10 mol %)/Me<sub>2</sub>Zn$ (1 equiv), 3 h at ambient temperature in THF/ether (1:2)] (Scheme 5). Two isomers **7a** and **b** (1:1.8 ratio) derived from the 1,4-addition of **1a** and the subsequent intramolecular aldol reaction of the enolate intermediate were obtained in 65% yield. It turned out that both isomers **7a**,**b** possess *cis* stereochemistry between the hydroxyl group and the phenyl ketone (Scheme 5).

Although the stereochemistry and the metal species of the enolate generated *in situ* is unclear, we could confirm that the intramolecular aldol reaction *did* proceed to provide *cis* stereochemistry.<sup>12</sup> Regarding the role of additives (Me<sub>2</sub>Zn or Me<sub>2</sub>AlCl) in the present transformation we must await further study.<sup>13</sup> In summary, we are able to show that an acylzirconocene chloride complex as an acyl group donor reacts with  $ω$ -unsaturated  $α, β$ -enone and -ynone under Pd-

<sup>(9)</sup> Ester instead of a ketone carbonyl in **3** failed to yield the corresponding product. The length of the tethering chain in **3** or **5** affected the reactivity; that is, a shorter or longer tethering chain did not give the cyclization product.

<sup>(10)</sup> The same type of transformations with Ti complex were reported. (a) Urabe, H.; Suzuki, K.; Sato, F. *J. Am. Chem. Soc*. **1997**, *119*, 10014. (b) Suzuki, K.; Urabe, H.; Sato, F. *J. Am. Chem. Soc*. **1996**, *118*, 8729.

<sup>(11)</sup> Recently, Montgomery et al. reported the formation of metallacycle through the reaction of alkynyl enals with Ni(0). Amarasinghe, K. K.; Chowdhury, S. K.; Heeg, M. J.; Montgomery, J. *Organometallics* **2001**, *20*, 370.

<sup>(12)</sup> An attempted trapping of the enolate intermediate with TMSCl gave a complex mixture of products, and we could not isolate the TMS-enol ether.

<sup>(13)</sup> We have examined other organozinc or organoaluminum reagents. However, only Me<sub>2</sub>Zn or Me<sub>2</sub>AlCl showed reasonable efficiency. The effect of organozinc has been discussed in the Ni-catalyzed reactions. See ref 3.





 $Me<sub>2</sub>Zn$  (Me<sub>2</sub>AlCl)-catalyzed conditions to give bicyclo<sup>[3.3.0]</sup> compounds. The stereoselective preparation of bicyclo[3.3.0] compounds is an important subject and excellent procedures

have been devised.14 We hope to demonstrate soon various means for utilizing our strategy to construct significant molecules.

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**Supporting Information Available:** General experimental procedure and characterization data for products reported in Tables 1 and 2 and Scheme 5 (<sup>1</sup>H NMR, <sup>13</sup>CNMR, IR, and HRMS). This material is available free of charge via the Internet at http://pubs.acs.org.

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<sup>(14) (</sup>a) Schore N. E. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Paquette, L. A., Eds; Pergamon: Oxford, UK, 1991; Vol. 5, p 1037. (b) Negishi, E. In *Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Paquette, L. A., Eds; Pergamon: Oxford, UK, 1991; Vol 5, p 1163 and the references therein.